

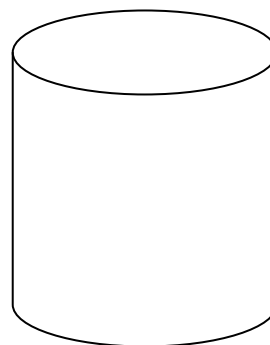
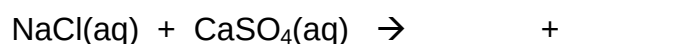
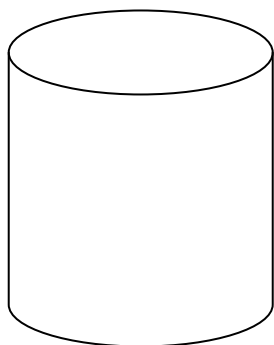
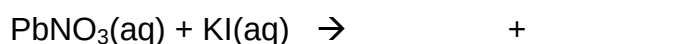
Double Replacement Reactions



Directions: Complete and balance the following reactions and determine if a precipitate or gas will be formed. Use your reference tables for help

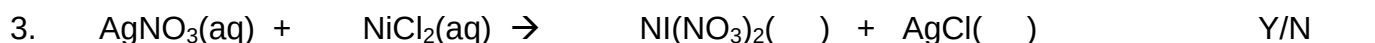
- In order for a reaction to occur there must be evidence of a chemical change
- We can predict if two compounds in aqueous solution will react by using the solubility guidelines in your Reference tables.
- Insoluble Means: _____
- Soluble Means: _____

For Example:



Directions: For each of the following predict if a reaction will occur using your solubility guidelines in your reference tables. Balance each equation and write the physical states in the spaces provided.

For Practice: _____ Will a reaction occur?





Will this Reaction occur? _____



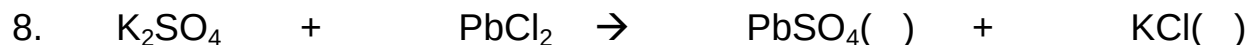
Will this Reaction occur? _____



Will this Reaction occur? _____



Will this Reaction occur? _____



Will this Reaction occur? _____



Will this Reaction occur? _____



Will this Reaction occur? _____

