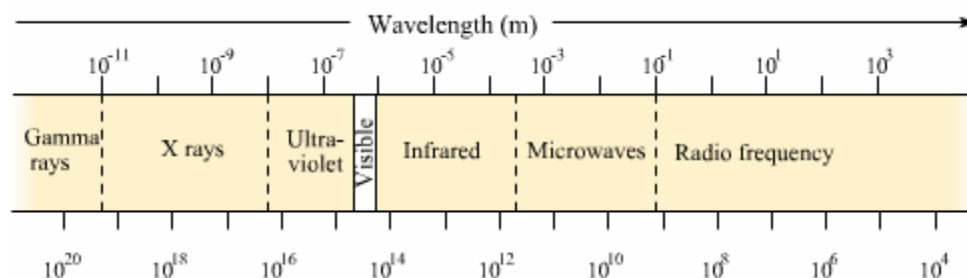


Chapter 6 Electronic Structure of Atoms

Equation Sheet

We will use the following equations in chapter 6



$$\nu\lambda = c \quad \text{or} \quad c = \lambda f$$

$$E = h\nu$$

$$E_n = (-R_H) \left(\frac{1}{n^2} \right) \quad \text{where } n = 1, 2, 3, 4, \dots$$

$$\Delta E = E_{\text{final}} - E_{\text{initial}} = h\nu$$

$$\nu = \frac{\Delta E}{h} = \frac{R_H}{h} \left(\frac{1}{n_i^2} - \frac{1}{n_f^2} \right)$$

$$\lambda = \frac{h}{m\nu}$$

$$Z_{\text{eff}} = Z - S$$

