

Name _____

Regents Chemistry

Naming Type II Binary Compounds

Directions: Complete the table below using your reference tables for help.

Chemical Formula	Cation Present	Anion Present	Cation Name	Anion Name	Compound Name
CuCl_2					
MnO_2					
PbCl_4					
CoBr_2					
CrCl_3					
PbBr_2					
PbBr_4					
FeS					
Fe_2S_3					
CoCl_3					

Directions: Write the chemical formula for each of the following names

Chemical Name	Compound Formula
Silver sulfide	
titanium (II) oxide	
titanium (IV) iodide	
copper (I) arsenide	
chromium (II) fluoride	
Tin (II) bromide	
Chromium (II) Oxide	
Chromium (III) Oxide	
Mercury (II) Iodide	
Cesium Selenide	
Calcium Phosphide	
Strontium Nitride	